## Acute and Chronic Toxicity of Alcohol Ethoxylates to the Green Alga, Desmodesmus (=Scenedesmus) subspicatus, and the Subsequent Development of Structure **Activity Relationships**

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Alcohol Ethoxylates (AE) are nonionic surfactants produced by the reaction of fatty alcohol and ethylene oxide and are common components of detergent formulations with approximately 1.1 million metric tons produced worldwide in 2004 (Hauthal 2004). AEs used in consumer product and industrial applications are actually complex mixtures of various compositions. AE molecules (also called ethoxymers) have the general formula CH<sub>3</sub>(CH<sub>2</sub>)<sub>n</sub>(OCH<sub>2</sub> CH<sub>2</sub>)<sub>v</sub>OH, where n is generally 11-15 and 17 and y is 0-18. Physico-chemical behavior is somewhat predictable as a function of the hydrophobic (alkyl) and hydrophilic (ethoxylate) moieties. Once AEs are discharged to the environment, these same properties affect sorption (Van Compernolle et al. 2005), biodegradation (Wind et al. 2005; Federle and Itrich 2005), and ecotoxicity (Boeije et al. 2005; Belanger et al. 2005).

Aquatic toxicity of AE mixtures has recently been explored by Boeije et al. (2005) who developed chronic mixture toxicity QSARs (quantitative structure activity relationship) for invertebrates (Daphnia magna) and fish (Pimephales promelas). The fundamental property linking AE homologue structure to toxicity is hydrophobicity (Roberts 1991). Because direct measurement of surfactant hydrophobicity by octanol/water partitioning is difficult due to their surface-active nature, log K<sub>ow</sub> can be well estimated mathematically by knowing molecular fragments and branching positions (Leo and Hansch 1979; Roberts 1991; Roberts and Marshall 1995). Thus, mathematically determined log Kow can be used as a hydrophobicity parameter when developing ecotoxicity QSARs for AE homologues (Boeije et al. 2005).

Almost all studies on AE toxicity have been performed on commercial mixtures. The presented research had the goal of determining the toxicity of pure AE homologues to algae that span a wide range of hydrophobicity in order to derive a QSAR for the log Kow-toxicity relationship. A standardized and consistent testing approach was taken to develop structure-activity-relationships for several common endpoints. These analyses would then allow ecotoxicity of AEs to algae to be properly incorporated in full-scale environmental risk assessments of complex AE mixtures.

## MATERIALS AND METHODS

Axenic cultures of the green alga *Desmodesmus* (= *Scenedesmus*) *subspicatus* were used throughout all tests. The source of the initial culture was from the Department of Plant Physiology (University of Goettingen / Germany).

Desmodesmus was cultured in mineral salt media according to Directive 92/69/EWG, Annex C3 corresponding to OECD 201 (OECD 1984). The culture medium was also used in performing toxicity tests.

Pure grade homogeneous AEs were purchased from Nikko Chemicals LtD. (Japan). Some physico-chemical characteristics are given in Table 1. Purity was confirmed by GC analysis (www.Nikkol.co.jp). Log  $K_{ow}$  for the test materials ranged from 3.45-7.29 based on the calculation methods of Leo and Hansch (1979) and Roberts (1991).

**Table 1.** Alcohol ethoxylates used in algal toxicity tests.

AE	Trade	Chemical name	Purity	Log	Solubility
	name		(%)	$K_{ow}$	(mM)
	(Lot#)				
$C_{10}EO_8$	BD-8SY	octaethylenglycol –	100	3.45	$4.59 \times 10^{-1}$
	(Lot 6003)	decylether			
$C_{12}EO_2$	BL-2SY	diethylenglycol –	100	5.13	$4.59 \times 10^{-3}$
	(Lot 1005)	dodecylether			
$C_{12}EO_4$	BL-4SY	tetraethylenglycol-	100	4.93	$7.33 \times 10^{-3}$
	(Lot 6003)	dodecylether			
$C_{12}EO_8$	BL-8SY	octaethylenglycol- 10		100 4.53	$2.24 \times 10^{-2}$
	(Lot 6015)	dodecylether			_
$C_{16}EO_2$	BC-2SY	diethylenglycol –	100	7.29	$1.00 \times 10^{-5}$
	(Lot 6001)	hexadecylether			_
$C_{16}EO_8$	BC-8SY	octaethylenglycol-	100	6.69	$5.35 \times 10^{-5}$
	(Lot 6001)	hexadecylether			

Standardized ecotoxicity test conditions followed those described in OECD Method 201 (OECD 1984) or EC Directive 92/69/EWG and are summarized in Table 2. Tests were begun when cells in culture were in exponential growth phase with initial cell densities of 1 x 10<sup>4</sup> cells/mL. Five to seven test concentrations were employed in each test with controls, all in triplicate. Test flasks (500mL) were held on a Clim-o-Shake® (horizontal shaker) at 100 rpm under constant illumination at approximately 2000 Lux. Cell density was confirmed at every time point (24, 48 and 72 h) by Coulter Counter, which was calibrated by manual cell counting in a Thoma-chamber before each counting routine.

Exposure concentrations were generally chosen based on experience and range finding tests. Dilution series were constructed using factors between 2 to 3 depending on the substance and test.

AE exposure concentrations were confirmed using the BiAS method (German Industry Norm 1980) in 2-3 treatments of each test. This method is specific for the major non-ionic surfactants of the EO type (>4EO) by analyzing bismuth after complexing the surfactant with barium and bismuth (BiAS=Bismuth Active Substance). Non-ionic surfactants <4EO are not quantitatively analyzed and therefore need to be adjusted by internal standards (eg. Nonylphenol EO<sub>10</sub>). The analytical limit of detection is around 5-10 $\mu$ g/L. Because analytical recovery of C<sub>12</sub>EO<sub>4</sub> dropped below 80% during the test, the effects in this test were based on the arithmetic mean of the nominal concentrations (e.g. 65.5%). Due to solubility limits the C<sub>16</sub>EO<sub>2</sub> AE was sonicated (ultrasound) in the test medium at nominal concentrations of 0.1-10 mg/L in order to enhance bioavailability. No additional carrier solvent was used. No analytical confirmation was possible for this test compound and effects were calculated on nominal concentrations.

**Table 2.** Algal ecotoxicity test conditions.

Parameter	Value		
Duration	72 h		
Light	2000 Lux, full spectrum		
Initial inoculum	$1 \times 10^4 \text{ cells/mL}$		
Counting method	Calibrated Coulter counter		
Temperature	23 +/- 2° C		
Shaker rotation	100 rpm		
Test medium	Mineral Salt (EU 1992)		
Frequency of counts	Every 24 h		

Several endpoints were evaluated based on cell count data and included population growth rate (0 to 72 h) and yield (area under the growth curve, 0 to 72 h) as described in standard protocols (OECD 1984; EU 1992).

Endpoints were statistically evaluated by exposure-response modeling (point estimate by regression) and threshold estimation (analysis of variance) techniques. Exposure-response modeling employed the non-linear, iterative regression technique of Bruce and Versteeg (1992). This technique has the advantage of reliably estimating  $EC_x$  values where x denotes a specified effect level between 1 and 99 while including the common non-monotonic trends observed at low exposure concentrations in algal tests (such as the hormetic response). Both 72-h  $EC_{50}$  and  $EC_{20}$  levels were determined to estimate acute and chronic responses, respectively. Thresholds for the no-observed-effect-concentration (NOEC) and lowest-observed-effect-concentration (LOEC) were concluded based on one-way analysis of variance (ANOVA) followed by an *a posteriori* Dunnet's multiple range test. In all cases significance was inferred at  $\alpha = 0.05$ . All statistical analyses were conducted in SAS (1999).

## RESULTS AND DISCUSSION

The six pure AE homologues varied in toxicity by approximately two orders of magnitude across the log  $K_{ow}$  range of 3.45-7.29 (Figure 1, Tables 3 and 4).  $C_{10}EO_8$  was the least toxic and  $C_{16}EO_8$  was the most toxic. In general, AE toxicity increases with increasing alkyl chain length and decreases with increasing ethoxylate chain length (Wong et al. 1997; Boeije et al. 2005). This is directly related to the influence of these factors on the hydrophobicity of the molecule.

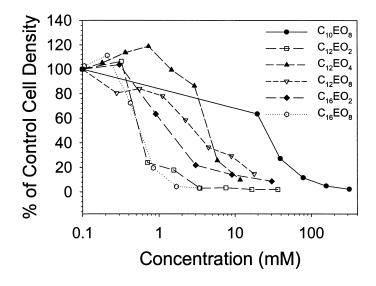


Figure 1. Response curves for all tests

**Table 3.** Acute 72-h EC<sub>50</sub> (in mM) estimations for various pure AE homologues.

AE	72-н рорг	lation growth rate	72-h yield or area under the		
				curve	
	$E_rC_{50}$	95% Lower-Upper	$E_bC_{50}$	95% Lower-Upper	
		Confidence Limits		Confidence Limits	
$C_{10}EO_8$	0.0859	0.0768-	0.0253	0.0186-	
		0.0958		0.0339	
$C_{12}EO_2$	0.00103	0.00645-	0.000653	0.000369-	
		0.001653		0.001167	
$C_{12}EO_4$	0.00704	0.00665-	0.00334	0.00296-	
		0.00751		0.00378	
$C_{12}EO_8$	0.0287	0.0251-	0.00450	0.00368-	
		0.0331		0.00551	
$C_{16}EO_2$	0.00900	0.00624-	0.00100	0.000545-	
		0.0130		0.00181	
$C_{16}EO_8$	0.000976	0.00657-	0.000539	0.000387-	
		0.01447		0.000791	

**Table 4.** Chronic 72-h EC<sub>20</sub> endpoints (in mM) for various pure AE homologues.

Table 4. Chrome 72-11 Ee20 endpoints (in mivi) for various pure AE nomologues.							
AE	72-h population growth rate			72-h yield or area under the			
					curve		
	$E_rC_{20}$	95% Conf	NOEC	$E_bC_{20}$	95% Conf	NOEC	
		Limits			Limits		
$C_{10}EO_8$	0.0282	0.0233-	0.0196	0.00902	0.00569-	0.0196	
		0.0341			0.01412		
$C_{12}EO_2$	0.00518	0.00255-	0.000321	0.000201	0.000088-	0.000279	
		0.01051			0.000471		
$C_{12}EO_4$	0.00318	0.00282-	0.000718	0.00174	0.00141-	0.000718	
		0.00356			0.00218		
$C_{12}EO_8$	0.00493	0.00405-	0.000279	0.00121	0.00084-	0.000321	
		0.00600			0.00173		
$C_{16}EO_2$	0.00133	0.000667-	0.000303	0.000273	0.00091-	0.000421	
		0.00276			0.00636		
$C_{16}EO_8$	0.000370	0.000185-	0.000421	0.000253	0.000135-	0.000303	
		0.000741			0.000438		
	·			•			

From estimations based on Kow/toxicity correlations,  $C_{16}EO_2$  should have been the most toxic AE. However, this AE is the most sorptive and least soluble homologue that was tested in this series. Analytical confirmation of exposure in the test also indicated  $C_{16}EO_2$  was the most variable with declining test material concentrations as time progressed (Table 5). It is quite likely that the AE was sorbed to algal cells and this effect increased with time as population growth occurred. Still, the test was successful in that exposure-response was evident and both  $E_rC_x$  and  $E_bC_x$  endpoints were reliably quantified.

**Table 5.** Analytical confirmation of selected exposure concentrations in AE toxicity tests

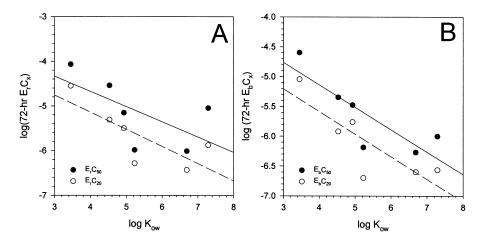
AE	Concentration	Percent of Nominal
	(mM)	$Mean \pm SD (n = 4)$
$C_{10}EO_8$	$7.84 \times 10^{-5}$	$99.4 \pm 1.0$
	$3.14 \times 10^{-5}$	$101.6 \pm 1/6$
$C_{12}EO_2$	$7.55 \times 10^{-7}$	$96.6 \pm 39.4$
	$3.65 \times 10^{-5}$	$105.0 \pm 34.1$
$C_{12}EO_4$	$4.42 \times 10^{-6}$	$65.5 \pm 24.7$
	$8.84 \times 10^{-6}$	$96.7 \pm 14.1$
$C_{12}EO_8$	$8.92 \times 10^{-6}$	$103.4 \pm 2.6$
	1.78 x 10 <sup>-5</sup>	$102.4 \pm 4.2$
$C_{16}EO_2$	$3.03 \times 10^{-6}$	$65.0 \pm 43.5$
	9.09 x 10 <sup>-6</sup>	$69.0 \pm 44.9$
$C_{16}EO_8$	$3.37 \times 10^{-6}$	$98.4 \pm 4.6$

Chronic toxicity was estimated as the  $EC_{20}$  and the mean ratio of the  $EC_{50}$ : $EC_{20}$  was 3.7 and 2.8 for the  $E_rC_x$  and  $E_bC_x$ , respectively. This is in the range historically observed for acute  $EC_{50}$ /chronic NOEC values for algae (van de

Plassche et al. 1999; ECETOC 2003) and provides support for the  $EC_{20}$  as a reasonable replacement for the no-observed-effect-concentration.

AE structure activity relationships were evaluated with log  $K_{ow}$  as the predictor variable (Table 6, Figure 2). The acute  $E_bC_{50}$  and chronic  $E_bC_{20}$  possessed the highest  $r^2$  values (approximately 0.7, p < 0.001). The slope of the QSARs were close to -0.4 which is very similar to those previously determined for other AE QSARs with daphnids (-0.5) and fish (-0.3) (Boeije et al. 2005). The first acute algae toxicity QSAR for pure AE-homologues was reported by Willing (2000). While the author related toxicity (expressed as the  $E_rC_{50}$ ) to the carbon chainlength and EO-moieties of AEs, the aquatic toxicity of AEs has been hypothesized to be driven by a non-polar narcotic mode of action and thus being related to log  $K_{ow}$  (Roberts 1991; Roberts and Marshall 1995). The  $r^2$  of the  $K_{ow}$ -based  $ErC_{50}$  QSAR in Table 6 is not as high as that of Willing (2000,  $r^2 = 0.994$ ).

were of the form. Log (Effect Endpoint in inivi) $-p_0 + p_1 + (\log R_{ow})$ .						
QSAR	Endpoint	Slope	Intercept	$\mathbf{r}^2$	p-value	
Acute	$E_rC_{50}$	-0.340	-3.315	0.586	0.003	
	$E_bC_{50}$	-0.375	-3.646	0.699	< 0.001	
Chronic	$E_rC_{20}$	-0.383	-3.610	0.609	< 0.001	
	$E_bC_{20}$	-0.378	-4.072	0.720	< 0.001	



**Figure 2.** 72-h  $E_rC_x$  QSARs for growth rate (A) and  $E_bC_x$  for yield (area under the curve) (B).

However, the latter had the statistical advantage of a two-term model with only two remaining degrees of freedom which perhaps biased the  $r^2$  value. Of the five studies, it is also relevant to note that the most hydrophobic AE that was sufficiently soluble to determine toxicity was  $C_{16}EO_2$ . The toxicity of this test

compound is much less than expected, perhaps indicating some solubility issues. If the data point is removed, the four QSARs above have  $\rm r^2$  values of 0.73-0.85. We have chosen not to cull values out of the data set as the tests were otherwise fairly well-behaved (such as good exposure-response profiles and low replicate variability). It is interesting to note that the slopes of all three taxonomic groups (fish, daphnia and algae) are similar and thus it can be hypothesized that the same mode of action applies. In addition, no one group is uniquely sensitive or insensitive as the intercepts also range between -0.3 to -0.4. Based on the data from this study, the  $\rm E_bC_{20}$  QSAR is recommended for use in the evaluation of AE distributions in the aquatic environment and subsequent risk assessment (Belanger et al. 2005). This relationship is summarized as:

$$Log(72-h E_bC_{20} in mM) = -0.378 * logK_{ow} - 4.072$$

There is a lack of published data on toxicity of AEs to algae. Historically, the interpretation of toxicity of AEs to algae was hampered due to a lack of analytical confirmation of exposure (Yamane et al. 1984; Lewis and Hamm 1986). In fact, all published data on algal toxicity is based on nominal exposures of commercial mixtures. Previous historical summaries of algal sensitivity to AEs indicated that algae were perhaps an order of magnitude less sensitive than invertebrates and fish (Talmadge 1994; van de Plassche et al. 1999). Based on the new data this appears not to be the case.

## **REFERENCES**

- Belanger SE, Dorn PB, Toy R, Boeije G, Marshall SJ, Wind T, Van Compernolle R, Zoeller D (in press) Aquatic risk assessment of alcohol ethoxylates in North America and Europe. Ecotoxicol Environ Saf.
- Boeije G, Cano ML, Marshall SJ, Belanger SE, Van Compernolle R, Dorn PB, Gumbel H, Toy R, Wind T (in press) Ecotoxicity QSARs for alcohol ethoxylate mixtures based on substance-specific toxicity predictions. Ecotoxicol Environ Saf
- Bruce RD, Versteeg DJ (1992) A statistical procedure for modeling continuous toxicity data. Environ Toxicol Chem 11:1485-1494
- ECETOC (2003) Aquatic hazard assessment II. (European Centre for Toxicology and Ecotoxicology of Chemicals) Technical Report No. 91. Brussels, Belgium. 165pp
- EU (European Union) (1992) Algal inhibition test. EU Method C.3. 92/69/EEC. Official Journal of the European Union L 383 A
- Federle TW, Itrich NR (in press) Fate of free and linear alcohol and alcohol ethoxylate and derived fatty alcohol in activated sludge. Ecotoxicol Environ Saf
- German Industry Norm DIN 38409 H 23 (1980) Deutsche Einheitsverfahren zur Wasser-, Abwasser- und Schlammuntersuchung Summarische

- Wirkungs- und Stoffkenngrößen (Gruppe H); Bestimmung der methylenblauaktiven und bismutaktiven Sustanzen.
- Hauthal HG (2004) CESIO 2004 Dynamic surfactants and nanostructured surfaces for an innovative industry. SÖFW; 130; 10: 3-17
- Leo AJ, Hansch C (1979) Substituent Constants for Correlation Analysis in Chemistry and Biology. Wiley & Sons, New York, New York
- Lewis MA, Hamm BG (1986) Environmental modification of the photosynthetic response of lake plankton to surfactants and significance to a laboratory-field comparison. Water Res 20:1575-1582
- OECD (1984) Alga Growth Inhibition Test, Method 201. Guideline for Testing of Chemicals, Organization for Economic Cooperation and Development, Paris, France 14pp
- Roberts DW (1991) QSAR issues in aquatic toxicity of surfactants. Sci Total Environ 109:557-568
- Roberts DW, Marshall SJ (1995) Application of hydrophobicity parameters to prediction of the acute aquatic toxicity of commercial surfactant mixtures. SAR QSAR Environ Res 4:167-176
- SAS (Statistical Analysis System) (1999) SAS/STAT User's Guide, Version 8.0. SAS Institute, Cary, NC
- Talmadge, SS (1994) Environmental and human safety of major surfactants: Alcohol ethoxylates and alkylphenol ethoxylates. Lewis Publishers, Boca Raton, Florida 374p
- Van Compernolle R, Cano ML, McAvoy DC, Sherren A, Belanger SE (in press)
  Development of a sorption QSAR for fatty alcohols and alcohol
  ethoxylates. Ecotoxicol Environ Saf
- van de Plassche EJ, de Bruijn JHM, Stephenson RR, Marshall SJ, Feijtel TCJ, Belanger SE (1999) Predicted no-effect concentrations and risk characterization of four surfactants: Linear alkyl benzene sulfonate, alcohol ethoxylates, alcohol ethoxylated sulfates, and soap. Environ Toxicol Chem 18:2653-2663
- Willing A (2000) Assessment of the ecological properties of various well known and new nonionic surfactants. Proceedings of the 5<sup>th</sup> CESIO World Surfactant Congress Firenze, Italy 2000
- Wind T, Stephenson RJ, Eadsforth CV, Sherren A, Toy R (in press)
  Determination of the fate of alcohol ethoxylate homologues in a laboratory
  continuous activated sludge unit study. Ecotoxicol Environ Saf
- Wong DCL, Dorn PB, Chai EY (1997) Acute toxicity and structure-activity relationships of nine alcohol ethoxylate surfactants to fathead minnow and *Daphnia magna*. Environ Toxicol Chem 16:1970-1976
- Yamane AN, Okada M, Sudo R (1984) The growth inhibition of planktonic algae due to surfactants used in washing agents. Water Res 18:1101-1105